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WORKSHEET Ionic Bonds Use with

Chapter 7, Section 7.2 1. How many

valence electrons does a neutral

magnesium (Mg) atom have? 2. What is

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the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl) atom have? 4.

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WORKSHEET The Activity Series Use
with Chapter 9, Section 9.2 1. For each of
the following pairs of elements, underline
the one that would replace the other
element in a compound.

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Answer to question 5, how many electrons can fit in the entire 5th energy level? can (A V el is a ha AC5a) 50 ÷ 7. Write the notation for an electron spinning clockwise in a p sublevel in the 4th ...

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WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there

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calcium and oxygen 2. Yes; each contains equal numbers of protons and electrons. 3. ionization energy 4. positive; cation 5. negative; anion 6. No; it is not a stable octet of electrons. 7.

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