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STUDY GUIDE FOR CONTENT\_MASTERY For each of the following chemical formulas, write the correct name of the ionic compound represented. You may refer to the periodic table on pages 156–157 and Table 8.7 for help. 19. NaI Sodium iodide 20. CaCl<sub>2</sub> 21. MgO 22. 23. LiHSO<sub>4</sub> 24. NH<sub>4</sub>Br 25. 26. KBrO<sub>3</sub> 28. 29. 30. 31. (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> 32. NaBrO<sub>3</sub> 33. 34. Fe(10 ) 32 Calcium Chloride

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