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process by which
materials give off
rays emitted by
uranium atoms.
radiation. the
penetrating rays
and particles

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emitted by a
radioactive source.
radioisotopes.

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section describes
the nature of
radioactivity and
the process of radio-
active decay. It
characterizes alpha,
beta, and gamma
radiation in terms

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process that
decreases the
number of slow
moving neutrons in
a nuclear reactor;
this is accomplished
by using control
rods made out of
material such as
cadmium, which

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absorbs neutrons.

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In your notebook,

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solve the following

problems. SECTION

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RADIATION 1.

What happens to

the mass number

and atomic number

of an atom that

undergoes beta

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decay? 2. A radioisotope of an element undergoes alpha particle decay. How do the atomic number and mass number of the particle change? 3.

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the nucleus atoms
contain electrons
protons and

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neutrons protons
and neutrons are in
the center of the
atom known as the
nucleus and
together they are
called nucleons all
of an atoms positive
charge and most of
the atoms mass is

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