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Doping of

Semiconductors The addition of a small percentage of foreign atoms in the regular crystal lattice of silicon or germanium produces dramatic changes in their electrical properties, producing n-type and p-type semiconductors.

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Extrinsic or Impure

Semiconductor: N -

Type Semiconductors:

Ge Ge Ge Ge Ge Ge Ge Ge

Ge + + As 0.045 eV E_g

= 0.74 eV C.B V.B

Donor level - When a semiconductor of Group IV (tetra valent) such as Si or Ge is doped with a penta valent impurity (Group V elements such as P, As or Sb), N type semiconductor is

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semiconductor, each

atom behaves as if there

are 8 electrons in its

valence shell and

therefore the entire

material behaves as an

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INTRODUCTION.

Semiconductors :-Most of the solids can be placed in one of the two classes: Metals and insulators. Metals are those through which electric charge can easily flow, while insulators are those through which electric charge is difficult to flow.

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