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E1-26 The room volume is  $(21 \times 13 \times 12) \text{ft}^3 = 32832 \text{ft}^3 = 928 \text{m}^3$ . The mass contained in the room is  $(928 \text{m}^3)(1.21 \text{kg/m}^3) = 112 \text{kg}$ ; E1-27 One mole of sugar cubes would have a volume of  $N_A \times 10^{-23} \text{cm}^3$ , where  $N_A$  is the Avogadro constant. Since the volume of a cube is equal to the length cubed,  $V = l^3$ , then  $l = \sqrt[3]{V} = \sqrt[3]{N_A \times 10^{-23} \text{cm}^3} = 2.15 \times 10^{-8} \text{cm}$ . E1-28 The number of seconds in a week is

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