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Graham ' s Law. • speed of diffusion of 2 different gases is inversely proportionality to the square root of their densities. Raoult ' s Law. • vapor pressure lowering due to effects of electrolytes or solutes in solution. • the partial vapor pressure of the solvent of an ideal solution is equal to the product of the mole fraction of the solvent and its vapor pressure in pure state.

Physical Pharmacy Notes | Intermolecular Force | Gases

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Dr Valizadeh (Teaching Materials) - Pharmacy faculty

Objectives of the lecture After completion of thislecture, the student should be able to: 1. Understand the various types of pharmaceutical solutions. 2. Define solubility, saturated & unsaturated solutions and polar & non polar solvents. 3. Describe complete & partial miscibility. 4.

Physical Pharmacy- lecture 1

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Physical Chemical and Biopharmaceutical Principal in the Pharmaceutical Science. Authors – Alfred N. Martin, Patrick j. sinko. Editor – Patrick j.sinko. Physical Pharmacy – Physical Pharmacy is the branch of pharmacy that concentrates on the application of physics and chemistry. It is the study of the effect dosage form has on their environment by addressing issues at the molecular level.

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Physical pharmacy Physical pharmacy. 1. PHYSICAL PHARMACY. 2. Matter: It is what the world is made of. 3. Kinetic Theory of Matter Matter is made up of particles which are in continual random motion. 4. Binding Forces between Molecules The Three States of Matter: In order for molecules to exist in aggregates in gases, liquids and solids Intermolecular forces must exist 4.

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• Physical properties - a characteristic that can be observed or measured without changing the identity or composition of the substance • Physical properties used to describe matter can be classified as: 1) Extensive – depends on the amount of matter in the sample - e.g. Mass, volume, length . 2) Intensive – depends on the type of

THE PROPERTIES AND STRUCTURE OF MATTER

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A concise guide providing the physicochemical background to the design and use of pharmaceutical dosage forms. This FASTtrack book is derived from the textbook Physicochemical Principles of Pharmacy and is designed to be used alongside it for those revision periods when time is short. It includes key points, tips, self assessment questions/answers and memory maps to aid with revision. For the new edition there will be an additional chapter on pharmaceutical nanotechnology.

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Publisher's Note: Products purchased from Third Party sellers are not guaranteed by the publisher for quality, authenticity, or access to any online entitlements included with the product. A complete practice-oriented introduction to physical pharmacy Written to clearly and simply explain how drugs work, this textbook explores the fundamental physicochemical attributes and processes important for understanding how a drug is transformed into a usable product that is administered to a patient to reach its pharmacological target, and then exists the body. Applied Physical Pharmacy, Third Edition begins with a review of the key biopharmaceutics concepts of drug liberation, absorption, distribution, metabolism, and excretion. These concepts, and others, set the framework for the subsequent chapters that describe physicochemical properties and process related to the fate of the drug. Other physical pharmacy topics important to drug formulation are discussed in the chapters that follow, which describe dispersal systems, interfacial phenomena, and rheology. The textbook concludes with an overview of the principles of kinetics that are important for understanding the rates at which many of the processes discussed in previous chapters occur. Chapters in this Third Edition retain the acclaimed learning aids of previous editions, including Learning Objectives, Practice Problems, Key Points, and Clinical Questions. In order to be of greater value to the pharmacy student, more clinical questions have been added, and many tables have been updated with more current products and excipients.

This monography is, in the first place, a commented guide that invites the reader to plunge into the thrilling world of zeta functions and their applications in physics. Different aspects of this field of knowledge are considered, as one can see specifically in the Table of Contents. The level of the book is elementary. It is intended for people with no or little knowledge of the subject. Everything is explained in full detail, in particular, the mathematical difficulties and tricky points, which too often constitute an insurmountable barrier for those who would have liked to become acquainted with that matter but never dared to ask (or did not manage to understand more complete, higher-level treatises). In this sense the present work is to be considered as a basic introduction and exercise collection for other books that have appeared recently. Concerning the physical applications of the method of zeta-function regularization here described, quite a big choice is presented. The reader must be warned, however, that I have not tried to explain the underlying physical theories in complete detail (since this is undoubtedly out of scope), but rather to illustrate - simply and clearly - the precise way the method must be applied. Sometimes zeta regularization is explicitly compared in the text with other procedures the reader is supposed to be more familiar with (such as cut-off or dimensional regularization).

This superb team of authors and editors provides a highly readable account of 'how the body works.' The book is structured around body systems and includes relevant diseases to illustrate the consequences of what happens when a part of a particular system fails. This book is aimed at students in biological sciences, medicine, dentistry, nutrition, pharmacy, physiotherapy and physical education. The book focuses on the core information required by students but further information is provided in small print for the interested or more advanced student.

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