

## Compounds Their Formulas Lab 7 Answers

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Report Sheet - Lab 7 D.2 Formulas of Ionic compounds and Ferrous Formula PO calcium ion iron O Negative Ion co 2- N 03  $\text{OH}^-$  - more Name Potassium carbonate Sodium nitrate Calcium bicarbonate Aluminum hydroxide Lithium hydroxide Potassium sulfate Positive Ion 24 D.3 Names of ionic compounds  $\text{CaSO}_4$   $\text{Al}(\text{NO}_3)_3$   $\text{Na}_2\text{CO}_3$   $\text{MgSO}_4$   $\text{Cu}(\text{OH})_2$  Calcium sulfate

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Compounds and Their Formulas Report Sheet Lab - 7 B.1 Physical properties Compound Appearance Density Melting Point B.2 Formulas of Ionic compounds Name Positive Ion Negative Ion Formula Sodium chloride Na + Cl- Magnesium chloride Calcium oxide Lithium phosphide Aluminum sulfide Calcium nitride B.3 Names of Ionic compounds K 2 S Potassium sulfide BaF 2 MgO Na 3 N AlCl 3 Mg 3 P 2 C. Ionic Compounds with Transition Metals C.1 Physical properties

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Compound name : Molecular weight: Molecular formula: 1: Acetic acid: 60.052 g/mol: CH 3 COOH: 2: Hydrochloric acid: 36.458 g/mol: HCl: 3: Sulfuric acid 98.072 g/mol: H 2 SO 4: 4: Acetate: 59.044 g/mol: CH 3 COO  $\ominus$  5: Ammonia: 17.031 g/mol: NH 3: 6: Nitric acid: 63.012 g/mol: HNO 3: 7: Phosphoric acid: 97.994 g/mol: H 3 PO 4: 8: Sodium phosphate: 119.976 g/mol: Na 3 PO 4: 9: Calcium carbonate: 100.086 g/mol: CaCO 3: 10: Ammonium sulfate

Chemical Compound Formula | Formula Chart  
lab 7 - Compounds and Their Formulas Report Sheet Lab 7 ... Question: Report Sheet-Lab 7 D.2 Formulas Of Ionic Compounds Formula Positive Ion Negative Ion Name Potassium Carbonate K Sodium Nitrate Calcium Bicarbonate Aluminum Hydroxide Lithium Phosphate Potassium Sulfate Co.2- D.3 Names Of Ionic Compounds  $\text{CaSO}_4$   $\text{Al}(\text{NO}_3)_3$   $\text{Na}_2\text{CO}_3$   $\text{MgSO}_4$

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