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Chemistry and Matter Chemistry
Central Science

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Chapter 1 Section 2 MATTER and Its Properties
Matter – Anything that has MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest

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unit of an element that still maintains the properties that element.

ELEMENT– a pure substance made up of only one type of atom.

MATTER and CHANGE Chapter 1 Section 1

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atomic radius and ionic radius. Circle
the letter of the choice that best
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the question. 1. Atomic radii cannot be measured ...

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Section 3.1 Properties of Matter pages
70–75 Problem-Solving Lab 1. Explain
why the flow of a compressed gas
must be controlled for practical and

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Table and Periodic Law Section 6.1

Development of the Modern Periodic

Table 1. octaves 2. eight 3. nine 4.

accepted 5. Dmitri Mendeleev 6.

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atomic mass 7. elements 8. atomic
number 9. Henry Moseley 10. protons
11. periodic law 12. properties 13. 14
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Chapter 5 25 Electrons For Atoms

Reviewing Vocabulary Match the definition in Column A with the term in Column B. Column A Column B

1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element
2. The minimum amount of energy that can be lost or

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Chapter Assessment

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Change. extensive property. intensive
property. extensive properties
(examples) intensive property
(examples) a property that depends on
the amount of matter in a sample. a

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property that depends on the type of matter in a sample not.... mass, volume, length, weight.

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Chapter 11 31 number of moles of
known substance C H C H C H O O O
CO CO CO H O H O H O number of
moles of unknown substance O CO H
O C H CO H O C H O H O C H O CO
use mole ratio 5 mol O 2 1 mol C 3 H
8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2
O 1 mol C 3 H 8 1 mol C 3 H 8 5 mol

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**MATH SKILLS TRANSPARENCY
MASTER 16**

A “chemical change” is a change that produces matter with a different composition than the original matter.
Chemical Change A change in which

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one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component elements.

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