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hammered or rolled into thin sheets is
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of Al = $131.6 \times 2 = 263.2 \text{ mol [1]}$ mass of
Al = $263.2 \times 26.98 = 7100 \text{ g [1]}$ b moles of
 $\text{Cr}_2\text{O}_3 = \dots$ mass of AgCl = 1.50×10^{-3}
 $\times 143.32 = 0.215 \text{ g [1]}$ c moles of $\text{AgNO}_3 =$
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