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a.) C₅H₁₂. b.) C₁₀H₂₂. c.) C₂H₂. 45. Ethanol (C₂H₅OH) and propane (C₃H₈) have similar molar masses. Does that mean they have the same boiling point? Explain. 46. For each of the pairs below, circle the substance with the higher boiling point: a.) I₂ and F₂. b.) HF and LiF. c.) CH₂O and CCl₄. 47.

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C 12.1 Pure substances and mixtures AQA GCSE Chemistry C12 Chemical Analysis Kerboodle Answers: Page No. 181. 1. Answer. In advertisement, a product added nothing to it is called as pure. For example: a pure orange juice without additives. While in chemistry- A pure substance is one that is made up of just one substance i.e either element or ...

New(9-1) AQA GCSE Chemistry C12 Chemical Analysis ...

Chapter 12 Part B1: Additional Alkyl and Aryl Radicals 2 points

Draw the structural or semi-structural formulas for all the alkyl/aryl radicals of the following. Place a X in place of the hydrogen which is removed to make the radical: (Not in Chapter 12-use the Internet)

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Chapter 12 – Introduction to Organic Chemistry: Hydrocarbons
Section 12.1 – Alkanes Goal: Identify properties characteristic of organic or inorganic compounds. Summary: Organic compounds: always contain carbon and hydrogen (though sometimes other nonmetals as well.) Inorganic compounds: all other compounds. Often ionic (metal + nonmetal)

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(12) disulfide bridges; salt bridges; H-bonding; dispersion forces; (metal-ion coordination is the 5th) (13) Cysteine (Cys) is the amino acid that can form disulfide bridges. (14) The acidic amino acids, aspartic acid (Asp) & glutamic acid (Glu) and the basic amino acids,

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lysine (Lys), arginine (Arg), and histidine (His) can form salt bridges because their side chains are ionized.

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