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Molar Mass, Mole Fraction,
Partial Pressure, Effusion
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~~Answers 10 — Gases: Part 1
of 12~~

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1 of 10 ~~Chapter 11 Gas Laws
Day 1 — Gases \u0026~~

Pressure Chapter 10 Gases

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*Intermolecular Forces and
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Partial Pressures \u0026

Vapor Pressure: Crash Course

Chemistry #15 Chapter 10 -

Gases Chapter 10 - Gases:

Part 2 of 12 Intermolecular

Forces Which gas equation do

I use? 11. Kinetic Theory of

Gases Part 5 General

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Pressure exerted by liquids

and gases chapter 11 class 8

science part 8

FSc Physics Book1, CH 11,

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Class 10th-Physics-Chapter

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Questions Chapter 11 Review

Gases Section

SECTION 1 Date CHAPTER 11

REVIEW Gases Class SHORT

ANSWER Answer the following

questions in the space

provided. b Pressure - orce

For a constant force, when

the surface area is tripled

the surface area pressure is

(a) doubled. as much. (c

ripled. 7-0 (d) unchanged.

Rank the following pressures

in increasing order. (c) 76

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Answers (a) 50 kPa O, 00ictbv-x

Home - Kenilworth Public Schools

If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because. A. there are more elastic collisions between the particles in a gas. B. gases are more compressible. C. the particles move faster in a gas and there is a greater distance between them.

*Chapter 11 Gases Review
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Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of

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Answers of gases. To describe calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This section describes how mixing gases affects the properties of the resulting mixture.

Chapter 11 - Gases

462 Chapter 11 Gases

Discovering the Relationships Between Properties If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider changes in the properties of the gases (pressure, volume, temperature, or number of

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Answers
gas particles) inside and outside the balloon.

Chapter 11 Gases - An Introduction to Chemistry

CHAPTER 11 REVIEW Gases

SECTION 1 SHORT ANSWER

Answer the following questions in the space provided.

1. b Pressure surface area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged.

2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m²

3.

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Gases: Section 1: Gases and
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*Chapter 11- Gases: Section
1: Gases and Pressure ...*

this theory explains some of
the properties of ideal
gases. In this chapter, you
will study the predictions
of kinetic-molecular theory
for gases in more detail.

This includes the
relationship among the
temperature, pressure,
volume, and amount of gas in
a sample. SECTION 11.1 Key

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Answers pressure newton
barometer millimeters of
mercury

*SECTION 11.1 Gases and
Pressure - Pickford High
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CHAPTER 11 REVIEW Gases

SECTION 2 SHORT ANSWER

Answer the following
questions in the space
provided. 1. State whether
the pressure of a fixed mass
of gas will increase,
decrease, or stay the same
in the following
circumstances: increase a.
temperature increases,
volume stays the same
decrease b. volume
increases, temperature stays
the same

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CHAPTER 11 REVIEW Gases
SECTION 1 SHORT ANSWER
Answer the following

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Answers in the space provided. 1. _____ Pressure = . For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. _____ Rank the following pressures in increasing order.

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