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Thermal,

Equilibrium When

the surroundings of

a system change,

the system itself

will change. After a

time, no further ma

croscopically-

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detectable changes
take place.

Transients have
died down. We say
the system is in
equilibrium. The
equilibrium state of
a system is
completely speci
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the systematic

study of

transformations of

matter and energy

in systems in terms

of a concept called

thermodynamic

equilibrium. The

word equilibrium

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implies a state of
balance.

Equilibrium
thermodynamics,
in origins, derives
from analysis of
the Carnot cycle.
Here, typically a
system, as cylinder
of gas, initially in
its own state of
internal
thermodynamic
equilibrium, is set

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out of balance via
heat input from a
combustion
reaction.

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axiomatic concept
of
thermodynamics. It
is an internal state

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system, or a
relation between
several
thermodynamic
systems connected
by more or less
permeable or
impermeable
walls. In
thermodynamic
equilibrium there
are no net

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